

無機化學部份，問答題共五題，每一小題 10 分，此部份總分為 50 分。

- Which of the following mixtures would be expected to have maximum boiling points and which to have minimum boiling points? Explain the reason.  
(a) Methyl acetate and chloroform, (b)  $C_6H_{12}$  and  $C_2H_5OH$ .
- Which of the following complexes obey the rule of 18 (EAN rule)?  
(a)  $Ni(NH_3)_6^{2+}$ ,  $Ni(CN)_4^{2-}$ ,  $Ni(CO)_4$  (b)  $Co(NH_3)_6^{3+}$ ,  $CoCl_4^{2-}$
- Give the approximate pKa values for the following acids:  
(a)  $H_3PO_3$ , (b)  $HNO_3$ , (c)  $HClO_4$ .
- The rate of reaction of  $O_2$  with  $trans-IrX(CO)(PPh_3)_2$  in benzene decreases in the order  $X = NO_2 > I > ONO_2 > Br > Cl > N_3 > F$ . Explain this observation.
- Sketch  $\pi$  bonding orbitals that result from combination of the following orbitals on separate atoms:  $p_x$  and  $p_x$ ,  $p_x$  and  $d_{xz}$ .

(背面仍有題目,請繼續作答)

分析化學部份，問答題共五題，每一小題 10 分，此部份總分為 50 分。

6. To prepare a solution of KHP (potassium hydrogen phthalate), you weigh out 5.103( $\pm 0.003$ )g and dissolve it in a volumetric flask whose volume is 250.00( $\pm 0.09$ ) mL. Express the molarity of the resulting solution, and its uncertainty, with the correct number of significant figures  
 Given: K:39.0983  $\pm$  0.0001 g/mol; O:15.9994  $\pm$  0.0003 g/mol;  
 H:1.00794  $\pm$  0.00007 g/mol; C:12.0107  $\pm$  0.0008 g/mol;
7. An organic compound weighing 5.714 mg produced 14.414 mg of CO<sub>2</sub> and 2.529 mg of H<sub>2</sub>O upon combustion. Find the weight percent of C and H in the sample.
8. Acid-base indicators are themselves acids or bases. Consider an indicator, HIn, which dissociates according to the equation  

$$\text{HIn} \rightleftharpoons \text{H}^+ + \text{In}^-$$
 Suppose that the molar absorptivity,  $\epsilon$ , is 2080 M<sup>-1</sup>cm<sup>-1</sup> for HIn and is 14200 M<sup>-1</sup>cm<sup>-1</sup> for In<sup>-</sup>, at a wavelength of 440 nm. The cell-length is 1.00 cm. A solution containing the indicator at a formal concentration of 1.84  $\times 10^{-4}$  M is adjusted to pH 6.23 and found to exhibit an absorbance of 0.868 at 440 nm. Calculate pK<sub>a</sub> for this indicator.
9. Is the complex of a metal ion with EDTA affected by pH of the solution?
10. How does the pH of pure water change with increasing temperature?