國立成功大學一○一學年度碩士班招生考試試題

系所組別: 資源工程學系乙組 考試科目: 熱力學

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编號:

- (1) Distinguish between expansion work against constant pressure and work of reversible expansion (10%)
- (2) What is (a) state function (b) exact differential (10%)
- (3) Justify Trouton's rule (10%)
- (4) Please express dG for a closed system in the absence of non-expansion work and at constant composition (10%)
- (5) What is Clausius-Clapeyron equation (10%)
- (6) Explain how colligative properties are used to determine molar mass (10%)
- (7) Explain what is meant by a regular solution (10%)
- (8) Define (a) phase (b) phase rule (c)component (10%)
- (9) Use the Ellingham diagram (Fig. at bottom) to identify the lowest temperature at which zinc oxide can be reduced to zinc metal by carbon (10%)
- (10) Distinguish between galvanic and electrolytic cells (10%)



(i) $M(s) + \frac{1}{2}O_2(g) \longrightarrow MO(s)$ (ii) $\frac{1}{2}C(s) + \frac{1}{2}O_2(g) \longrightarrow \frac{1}{2}CO_2(g)$ (iii) $C(s) + \frac{1}{2}O_2(g) \longrightarrow CO(g)$ (iv) $CO(g) + \frac{1}{2}O_2(g) \longrightarrow CO_2(g)$