## 國立成功大學七十九學年度確治及材料 考試(物理治全 試題) 第 1 頁

### 咨題要領:

- (1)請按題号順序回答
- (2) 答題語文僅限中文、英文及日文。

請設法簡明的説明下列問題:

- 1) 説明亜共晶組成合金之凝固過程:
  - a)凝固曲線特徵
  - b)組織形成過程及形態
  - c) 成分固定時、請列挙能影響機械性質的組織学要因

(本題10分)

2) 軟鋼在経過種種不同塑性加工及無処理之後,設肥粒鉄平均粒径自50 um 変化至20 um, 請定量的計算降伏点強度変化量 (Hall-Petch 之関係係数為k=2.5 kgf/mm<sup>-3</sup>2)

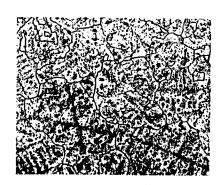
(本題10分)

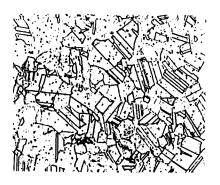
- 3)請用核生成及核成長的観点説明下列現象之理由
  - a) Martensite在安定相之体心立方鋼試料与Feg C上不生成,但却在過飽和之面心立方鋼試料上生成、為何?
  - b) 低温回火加熱処理時不析出FegC. 但却析出 E-- Carbide. 為何?

(本題 a, b各10分)

- 4) 有多数的合金 (例如 Fe Mn Al合金) 有下述現象存在, 請具体説明 理由:
  - -- 実施高温退火処理後、鋳造試料無退火双晶 (annealing twin) 出現而圧延試料却有多数的退火双晶出現 (参考下図), 為何?

(本題10分)





## 國立成功大學 + t 是年度 强党的名所考試( 物 理 治 全 試題) 并 5 月

(5.) The diffusion rate of carbon in d iron (ferrite)

and Y iron ( austenite) are given by

$$D_d = 0.0079 \text{ cm}^2/\text{sec } e^{(-18,100 \text{ cal/mole})/RT}$$

$$D_r = 0.21$$
  $cm^2/sec$   $e^{(-33,800 cal/mole)/RT}$ 

- (a) Calculate the two diffusion coefficients at 800°C and at 1000°C
- (b) Explain the magnitude of Dx compared to Dr in terms of atomic structure
- (c) Draw a composition-distance plot of a carbon-iron diffusion couple heat-treated at 800°C. (see iron-carbon phase diagram as below)
- (d) Why are commercial carburizing treatments of steel carried out when the steel is austenitic?

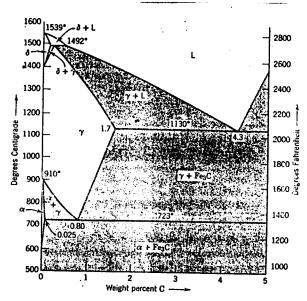
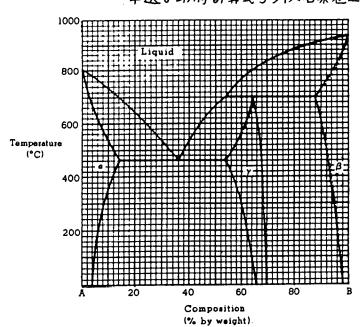


Figure . The phase diagram for metastable Fe-Fe<sub>3</sub>C.

#### 治及材料就( 國立成功大學 七代學年度 强为及行列 为世治至試題) Ą

Questions 1-10 relate to the phase diagram shown below. A and B are pure elements and α, β, γ are solid solutions. Assume equilibrium conditions unless otherwise specified. All temperatures are in degrees centigrade and all compositions are in per cent by weight. 單键。請將計算式子列入答案卷上。 (6) 佔10分



- Which of the following is the name of the isothermal reaction taking place at 700°C?
  - (A) Peritectic (B) Eutectic (C) Syntectic (D) Peritectoid (E) Eutectoid
- 2. The maximum solubility of A in B in per cent by weight is most nearly
  - (A) 2% (B) 4% (C) 12% (D) 14% (E) 36%
- 3. The amount of  $\gamma$  contained in a 120-kilogram alloy of 54 per cent by weight B at  $600^{\circ}$  C is most nearly (A) 14.1 kg. (B) 41.2 kg. (C) 49.4 kg. (D) 58.8 kg. (E) 70.6 kg.
- 4. If an alloy of 10 per cent by weight B is cooled from 440<sup>b</sup> C under nonequilibrium conditions, which of the following conditions will be present to the alloy 2 in the alloy?

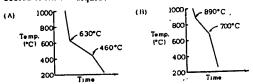
  - (A) Liquid immiscibility
    (B) Supersaturation
    (C) Pearlite formation
    (D) Martensitic transformation
    (E) Liquation at grain boundaries

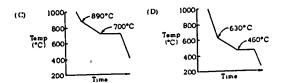
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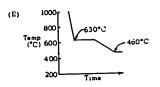
#### 國立成功大學 ++九學年度確沒及材料考試( 試題) 43 键 治金

- For an alloy of 70 per cent by weight A, the percentage of phases present at 250°C are most nearly
  - (A) 25.0% α and 75.0% β (B) 42.3% α and 57.7% γ (C) 57.7% α and 42.3% γ (D) 7.2% β and 92.8% γ (E) 92.8% β and 7.2% γ
- 6. The compositions of the phases present in an alloy of 80 per cent by weight B at 600° C are most nearly
  - (A)  $\alpha = 7\%$  B and  $\beta = 90\%$  B (B)  $\alpha = 7\%$  B and  $\gamma = 23\%$  B (C)  $\beta = 60\%$  B and  $\gamma = 40\%$  B (D)  $\beta = 90\%$  B and  $\gamma = 61\%$  B (E)  $\beta = 90\%$  B and  $\gamma = 65\%$  B
- If one assumes that pressure is constant at 1 atmosphere, what are the number of degrees of freedom for a system with a composition of 45 per cent by weight B at 720°C?
  - (A) 0 (B) 1 (C) 2 (D) 3 (E) 4
- If one assumes that A and B are ceramic oxides, the highest safe temperature at which a refractory brick made from 70 per cent by weight B could be used is most nearly
  - (A)  $440^{\circ}$  C (B)  $680^{\circ}$  C (C)  $780^{\circ}$  C (D)  $920^{\circ}$  C (E)  $1000^{\circ}$  C

9. Which of the following graphs best represents a cooling curve for an alloy of 20 per cent by weight B that is cooled from the liquid?







- 10. If the density of A is 2.25 grams per cubic centimeter and that of  $\gamma$  is 2.70 grams per cubic centimeter, the volume percentages of phases present for a 40 per cent by weight of B alloy at  $100^{\circ}$  C are most nearly
  - (A) 6.1% α and 93.9% γ (B) 35.4% α and 64.6% γ (C) 39.7% α and 60.3% γ (D) 44.1% α and 55.9% γ (E) 60.3% α and 39.7% γ

# 國立成功大學 + 4 學年度 遵治及村科考試(物理 冶金 試題) 第 5 页

- (7.) 請比較一般析出(precipitation)和離相分解(spinodal decomposition 站5分 或叫作節點分解)之不同。 說明請以簡要為原則。
- (8) 請由原子排列和鏈接的觀點簡答下列問題:
  - (a) 怎何一般金屬材料比陶瓷材料之延展性高?
  - (b) 温度對於一般金屬材料和陶瓷材料,導電性質的影響分別為何?
  - (c) 温度對於一般 金屬材料和陶瓷材料 "導熱"性質的影響分別為何?
- (9) 請簡要地解釋下列各名詞
  - (a) Anisotropy
  - (b) Bragg Law
  - (c) Born theory of ionic crystals
  - (d) Miscibility Gap (可用簡圖説明)
  - (e) Critical Stress Intensity Factor