編號:

177

國立成功大學102學年度碩士班招生考試試題

共 之頁,第一頁

系所組別: 電機工程學系甲組

考試科目: 電子材料概論

考試日期:0223, 節次:2

※ 考生請注意:本試題不可使用計算機

1. Foe the BCC unit cell, determine for the relationship between the cube side length a and the atomic radius R. Calculate the packing factor (20%)

- 2. Consider a simple cubic unit cell, side length a, with eight identical atoms, radius R, at the corner of a cube. There is one interstitial site at the center of cube. Calculate the radius of the interstitial site. (10%)
- 3. Look at the following section of the periodic table (10%)

II	III	IV	V	VI
Cd	In	Sn	Sb	Te

- a). Consider the compound semiconductor CdTe. If Sn impurities are inducted into CdTe and sit on Cd sites, wheat semiconductor type would result? Explain
- b). Consider the compound semiconductor InSb. In growing this semiconductor, excess indium atoms are introduced. The excess indium atoms sit on Sb sites. Explain which semiconductor type would result.
- 4. Please address the effect of temperature and impurity on the conductivity for metal, insulator and semiconductor, respectively.(20%)
- 5. Consider a piece of silicon doped with donor and acceptor impurities, with concentration N_D and N_A, respectively. Recall that, at a given temperature, the relationship np=constant always holds. In addition, not all donors and acceptors are ionized. Based on the principle of electrical neutrality, write down a relationship among the following four quantities: n, p, N_D⁺ (ionized donors), and N_A⁻ (ionized acceptors) (20%)
- 6. Refer to the Pb-Sn phase diagram as shown in Fig. (1)(20%)
 - a). State the composition of the eutectic alloy and the eutectic temperature.
 - b). What will happen to the eutectic alloy when the temperature is raised above the eutectic temperature?
 - c). Cool 100g of the eutectic alloy to slightly below the eutectic temperature. Calculate the amount of α and β phase at thermal equilibrium?
 - d). Take a Pb-Sn alloy with 25 wt% Sn. As we heat the alloy from room temperature, what is the temperature when melting starts?

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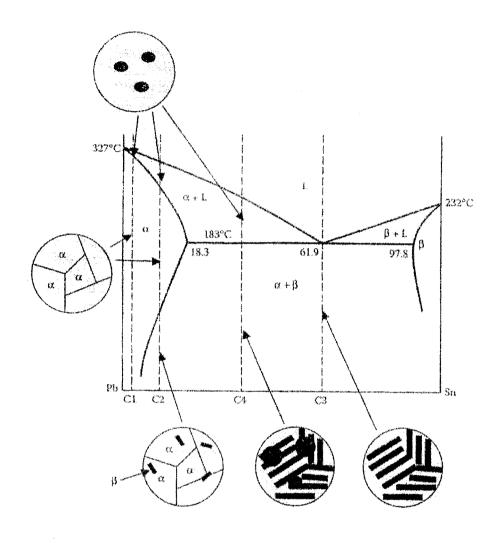
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Fig.(1)



Pb-Sn phase diagram