編號: 177

國立成功大學 105 學年度碩士班招生考試試題

系 所:電機工程學系 考試科目:電子材料概論

考試日期:0227,節次:2

第1頁,共1頁

- ※ 考生請注意:本試題不可使用計算機。 請於答案卷(卡)作答,於本試題紙上作答者,不予計分。
- 1 · Consider the FCC unit cell of the copper crystal (20%)
 - a. How many atoms are there per unit cell?
 - b. If R is the radius of the Cu atom, show that the lattice parameter is given by a = $2\sqrt{2}$ R
 - c. Calculate the atomic packing factor. For the FCC unit, assuming the atoms to be hard spheres.
 - d. Calculate the atomic concentration (number of atoms per unit volume) in Cu and the density of the crystal given that the atomic mass of Cu is 63.55 g/mol and the radius of the Cu atom is 0.128nm.
- 2. What are the energies required to ionize H atoms to H⁺ ions and further ionize He⁺ ions to He⁺⁺, respectively? (10%)
- 3. Find the resistance of a 1 cm³ pure silicon crystal. What is the resistance when the crystal is doped with arsenic if the doping is 1 in 10^9 , that is 1ppb? Atomic concentration in silicon is $5 \times 10^{22} \text{cm}^{-3}$, $n_i = 1 \times 10^{-10} \text{cm}^{-3}$, $\mu_e = 1350 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$, and $\mu_h = 450 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$. (15%)
- 4. Please address what is Hall Effect and show that if R_H is Hall coefficient and σ is the conductivity of a metal, then the drift mobility of the conduction electrons is given by $\mu_d = \sigma \times R_H (10\%)$
- 5. An n-type Si semiconductor containing 10^{16} phosphorus (donor) atom cm⁻³ has been doped with 10^{17} boron atoms cm⁻³. Calculate the electron and hole concentration in this semiconductor. (n_i =1x10¹⁰cm⁻³) (15%)
- 6. Draw the following direction vectors in cubic unit cells. (10%) [100], [110], [112], [-110], [-32-1]
- 7. Explain the following noun. (20%)
 - a). Frenkel defect
 - b). Eutectic point
 - c). Mean free path
 - d).Lattice-scattering-limited drift mobility
 - e). Impurity-scattering-limited drift mobility
 - f).Bragg's Law
 - g). Work function
 - h). Linear combination of atomic orbitals
 - i).Fermi energy
 - j).Schottky effect