

國立成功大學

113學年度碩士班招生考試試題

編 號：267

系 所：臨床藥學與藥物科技研究所

科 目：有機化學

日 期：0202

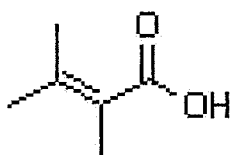
節 次：第 1 節

備 註：不可使用計算機

※ 考生請注意：本試題不可使用計算機。請於答案卷(卡)作答，於本試題紙上作答者，不予計分。

1. Multiple-Choice Questions (each 2%, total 10 %)

(1) Which of the following condensed formulas correctly represents the line-angle structure shown below?

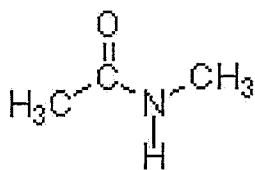


- A) $\text{CH}(\text{CH}_3)_2\text{CH}(\text{CH}_3)\text{CO}_2\text{H}$
 B) $\text{C}_2(\text{CH}_3)_3\text{CO}_2\text{H}$
 C) $(\text{CH}_3)_2\text{CC}(\text{CH}_3)\text{CO}_2\text{H}$
 D) $\text{C}(\text{CH}_3)_2\text{C}(\text{CH}_3)\text{CH}_2\text{CO}_2\text{H}$

(2) A sample of compound X is subjected to elemental analysis and the following percentages by weight are found: 39.97% C, 6.73% H, and 53.30% O. The molecular weight of X is 90. What is the empirical formula of X?

- A) C_6HO_8
 B) $\text{C}_2\text{H}_4\text{O}_2$
 C) $\text{C}_4\text{H}_{10}\text{O}_2$
 D) $\text{C}_3\text{H}_6\text{O}_3$
 E) CH_2O

(3) In the structure below, the sigma bond of the carbonyl is formed from the overlap of a(n) _____ atomic orbital of carbon and a(n) _____ atomic orbital of oxygen.



- A) sp, sp^2
 B) sp^3, sp^2
 C) sp^2, sp^2
 D) p, p

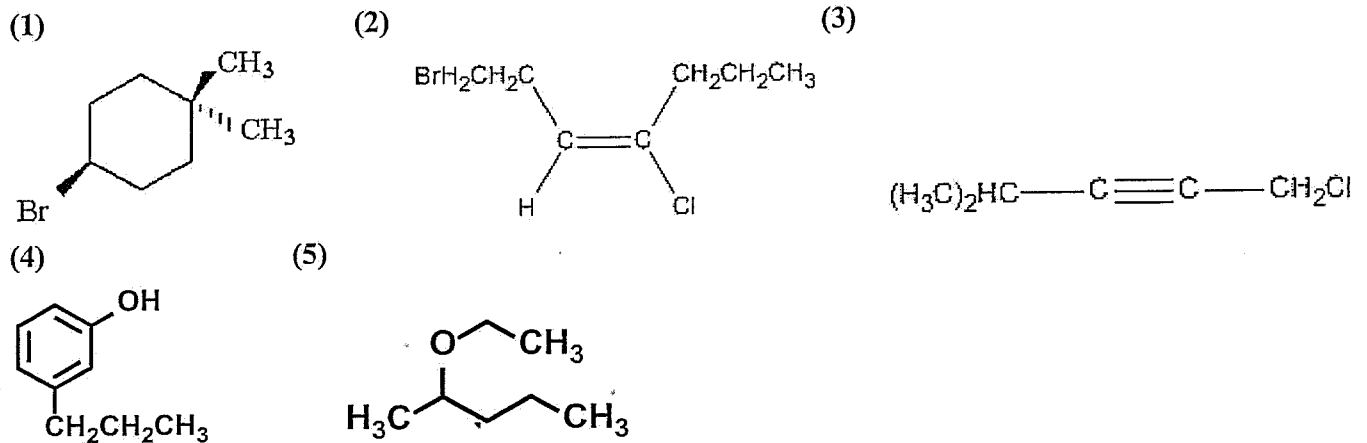
(4) How many secondary (2°) carbons are found in 5-ethyl-3,3,4-trimethylheptane?

- A) 1
 B) 4
 C) 2
 D) 3
 E) 6

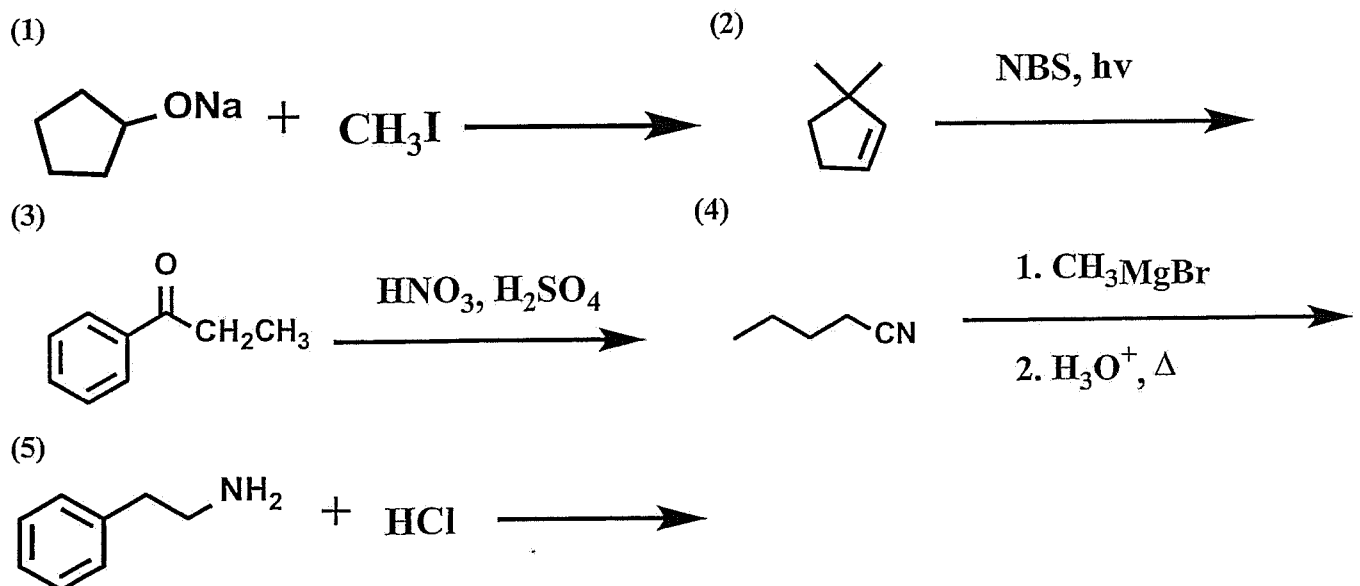
(5) Which of the following is not a possible termination step in the free radical chlorination of methane?

- A) $\cdot\text{CH}_3 + \text{Cl}_2 \rightarrow \text{CH}_3\text{Cl} + \text{Cl}\cdot$
- B) $\cdot\text{CH}_3 + \text{Cl}\cdot \rightarrow \text{CH}_3\text{Cl}$
- C) $\cdot\text{CH}_3 + \cdot\text{CH}_3 \rightarrow \text{CH}_3\text{CH}_3$
- D) $\cdot\text{CH}_3 + \text{wall} \rightarrow \text{CH}_3\text{-wall}$
- E) $\text{Cl}\cdot + \text{wall} \rightarrow \text{Cl-wall}$

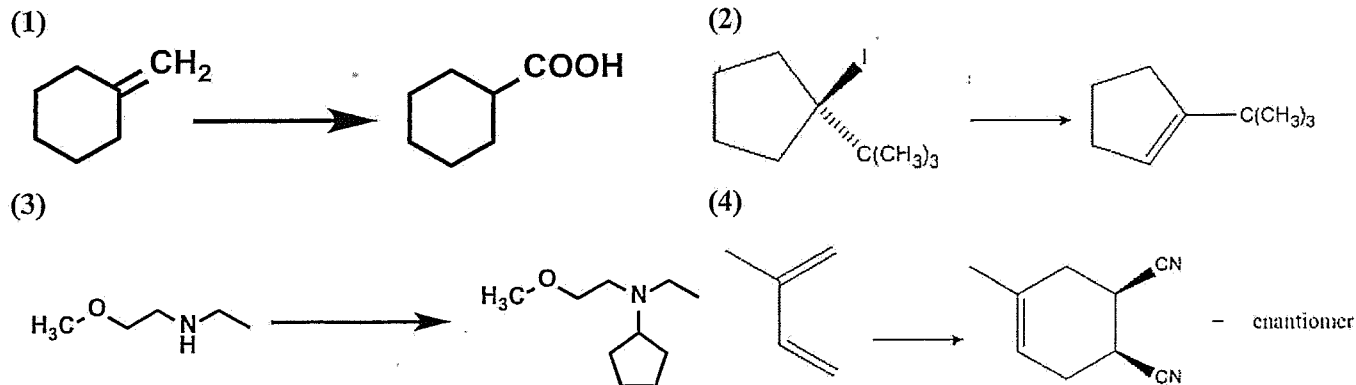
2. Assign the IUPAC names for the following compounds. (each 2%, total 10 %)



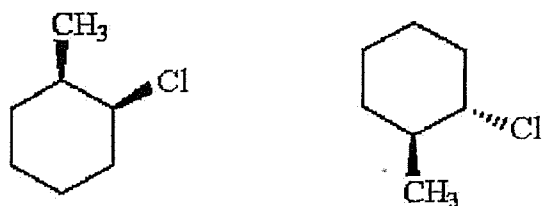
3. Complete the following reactions. (each 2%, total 10%)



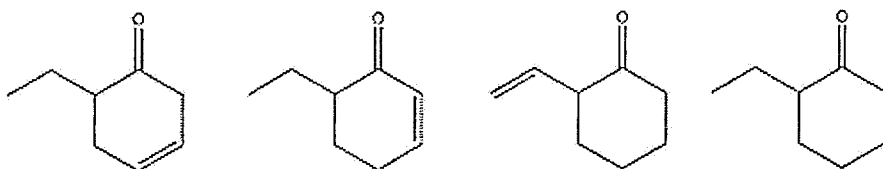
4. Complete the following multiple-step transformation. (each 5%, total 20%)



5. Which of the following terms best describes the pair of compounds shown: enantiomers, diastereomers, or the same compound? (5%)

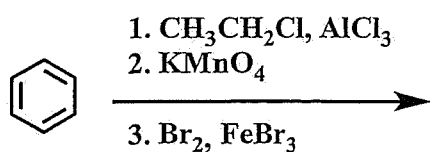


6. Which of the following compounds has the lowest carbonyl stretching frequency? (5%)



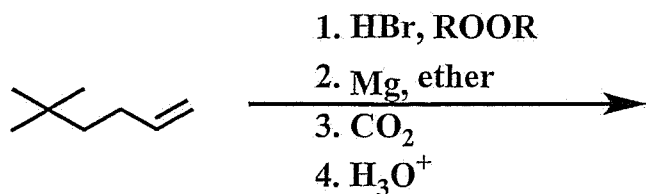
7. Compound I has a molecular formula of C_7H_{16} . In ^{13}C NMR, compound I gave 3 peaks and in 1H NMR it also gave 3 peaks, a doublet, a triplet and a multiplet. Provide a structure for compound I. (5%)

8. Provide the major organic product(s) of the reaction shown below. (5%)



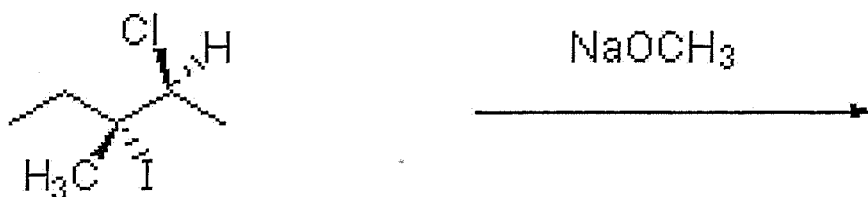
9. Provide a detailed, stepwise mechanism for the reaction of acetyl chloride with *n*-propylamine. (5%)

10. Provide the major organic product of the reaction shown below. (5%)

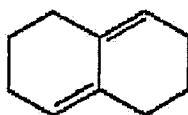


11. Draw the Newman projection of the highest energy conformation that results from rotation about the C2-C3 bond of 2-methylbutane. (5%)

12. Provide a structure for the major substitution and major elimination product resulting from the reaction below. (5%)



13. Provide the structure of the major product which results from 1,4-addition of Br₂ to the diene shown below. (5%)



14. Explain while this reaction won't proceed as it is written in the forward direction. (5%)

