Part I: 50%

- (一) 選擇題 (2% of each answer)
- When [S] = 2 K_M for an enzyme obeying classical Michaelis-Menten kinetics, the initial velocity, v_o, equals
 (a) Vmax (b) 0.67 Vmax (c) 0.5 Vmax (d) 0.33 Vmax
- When yeast extracts act on [1-14C] glucose, the radioactive label will end up in (a) [14C] CO₂ (b) [1-14C] ethanol (c) [2-14C] ethanol
 (b) (d) [14C] CO₂ and [1-14C] ethanol (e) [14C] CO₂ and [2-14C] ethanol
- 3. The PEP carboxykinase-catalyzed reaction requires(a) biotin (b) biotin and thiamine pyrophosphate (c) biotin and GTP(d) GTP (e) GTP and ATP
- 4. The only cytochrome in the respiratory chain that can transfer electrons to molecular oxygen is
 (a) cytochrome a (b) cytochrome a₃ (c) cytochrome b (d) cytochrome c
 - (e) cytochrome c₁
- Okazaki fragments will accumulate when a defect exists in(a) RNase H (b) primase (c) helicase (d) DNA gyrase (e) DNA ligase
- 6. The eukaryotic enzyme that is responsible for the formation of pre-mRNA is called
 - (a) RNA polymerase I (b) RNA polymerase II (c) RNA polymerase III (d) reverse transcriptase (e) telomerase
- A cell-free protein synthetic system requires each of the following except
 (a) ribosomes (b) mRNA (c) ATP and GTP (d) aminoacyl-tRNAs (e) DNA
- 8. 3-Phosphohydroxypyruate is an intermediate in the synthesis of (a) serine (b) alanine (c) threonine (d) aspartate (e) glutamine
- 9. Humans can synthesize each of the following amino acids from a citric acid cycle or glycolytic intermediate except
 - (a) alanine (b) aspartate (c) glutamate (d) threonine (e) serine

- 10. The glycolytic pathway requires each of the following except
 (a) NAD⁺ (b) ATP (c) ADP (d) biotin (e) phosphate
- (二) 填充題 (2% of each answer)
- 1. Please give two basic amino acids: (1) (2)
- 2. Please give two acidic amino acids: (3) (4)
- 3. The aminio acid that carries the ammonium ion from muscle to liver is (5)
- 4. The complete degradation of stearoyl –CoA through the β-oxidation pathway produces (6) FADH2, (7) NADH, and (8) acetyl-CoA molecules.
- 5. Oligo-dT column can be used to enrich which class of RNA (9)
- 6. The Vitamin (10) is important for maturation of collagen and required for the integrity of blood vessels.
- (三)簡答題 (5% of each answer)
- 1. Nucleosome and linker DNA
- 2. TCA (Tricarboxylic acid) cycle

Part II: 50%

- I. Multiple Choice (Choose only one correct answer) (2-points each)
- 1. Noncovalent bonds include all the following except:
 - a. A carbon-carbon double bond.
 - b. An ionic bond.
 - c. A hydrogen bond.
 - d. A van der Waals interaction.
- 2. Biomembranes are stabilized by all the following interactions except:
 - a. van der Waals interactions between the hydrophobic side chains of phospho-lipids.
 - b. Ionic bonds that stabilize the interaction of the phospholipid polar heard groups with water.
 - c. Hydrogen bonds that stabilize the interaction of the phospholipid polar head groups with water.
 - d. Carbon-carbon bonds between adjacent phospholipids.

- 3. All enzymes alter the rate of a chemical reaction through the following, except:
 - a. Forming a covalent bond with the substrate.
 - b. Binding to the substrate.
 - c. Bringing multiple substrates close to one another.
 - d. Raising the activation energy for formation of a reaction intermediate.
- 4. The rate of migration of a protein through an SDA-polyacrylamide get is *not* influenced by
 - a. Size of the protein.
 - b. Charge of the protein.
 - c. Pore size of the gel.
 - d. Strength of the electric field.
- 5. Which of the following occurs when the eukaryotic translational machinery encounters the TAG codon:
 - a. The bound preinitiation complex stops scanning and positions the Met- $tRNA_i^{met}$ at this site.
 - b. The termination factors recognize this codon and translation ends.
 - c. This codon is recognized by the corresponding anticodon of an empty tRNA molecule that is not linked to an amino acid.
 - d. This codon is not recognized by any factor that ultimately cause the translational machinery to stop.
- 6. An investigator would be able to distinguish a solution containing RNA from one containing DNA by
 - a. Heating the solutions to 82.5° C and measuring the absorption of light at 260 nm.
 - b. Comparing the T_m of each solution.
 - c. Monitoring the change in absorption of light at 260 nm while elevating the temperature.
 - d. Measuring the absorption of light at 260 nm.
- 7. The percentage of G.C base pairs in a DNA molecule is related to the T_{m} because
 - a. The stability of G.C and A.T base pairs is intrinsically different.
 - b. A.T base pairs require a higher temperature for denaturation.
 - c. The triple bonds of G.C base pairs are less stable than the double bonds of A.T base pairs.
 - d. The G.C content equals the A.T content.
- 8. The observation that plasama membrane proteins mix after cell fusion provides evidence for
 - a. Rotational movement of plasma membrane proteins.
 - b. The bilayer structure of biomembranes.
 - c. The fluid mosaic model.
 - d. Interaction between the plasma membrane proteins from two different cell types.

(背面仍有題目,請繼續作答)

- 9. Special structures called telomeres are needed in eukaryotic cells but not bacteria because
 - a. Eukaryotic cells contain liner chromosomes.
 - b. Eukaryotic cells have more than one chromosome.
 - c. Eukaryotic cells contain a nucleus.
 - d. Eukaryotic cells contain more forms of DNA polymerase.
- 10. Which of the following enzymatic activities does not play a role in mismatch repair:
 - a. Helicase.
 - b. Single-stranded exonuclease.
 - c. DNA ligase.
 - d. Primase.

II. Matching questions (2 points each)

Following is a list of hereditary metabolic defects involving loss of single enzymes of catabolism, and a second list of possible consequences of such defects. Match each enzyme defect with its *most likely* consequence (only one) from the second list.

- 1. Lack of pyridoxal kinase (catalyzes conversion of pyridoxal to pyridoxal phosphate)
- 2. Lack of isocitrate dehydrogenase
- 3. Lack of phosphofructokinase
- 4. Lack of glycogen phosphorylase kinase
- Lack of phosphorylase a phosphatase

Consequences:

- a. A lower than normal steady-state level of glycogen.
- b. Inability to use glycogen as an energy source, with no effect on ability to use galactose.
- c. Lethal; prevents use of carbohydrates for ATP production.
- d. Lethal; prevents complete oxidation of all fuel molecules.
- e. Inability to synthesize or degrade almost all amino acids.

III. Short Assays (5 points each)

- 1. What is the P/O ratio of oxidative phosphorylation?
- 2. Mutations that alter DNA can alter the function of an expressed protein. A mutation is characterized in a gene from a model organism and found to contain a single base change relative to the wild-type gene, yet the encoded protein still functions in this organism. How is this possible?
- 3. What is the main feature of allosteric control that makes it such a tremendously important concept?
- 4. How does insulin control the rate of glucose entry into fat cells?