國立成功大學七七學年度等物化學考試(有機化學試題)并6月

考生注意事项:所有考题移必在答案卷上作答。凡在問題卷上作答者無效。

-.是非題(每題 1分,T-TRUE or F-FALSE,選擇正確答案,答錯例如0.5分

- T F (1) Electron-releasing groups bonded to an amine nitrogen increase its basicity.
- T F (2) An N-alkyl sulfonamide is insoluble in sodium hydroxide because it has no acidic hydrogen.
- T F (3) In the uv-vis region, the intensity of absorption is independent of the number of absorbing molecules in the light path.
- T F (4) The most useful transition in the uv-vis regions are $n+\pi^*$ and $\pi+\pi^*$ transitions.
- T F (5) The D-series of aldohexoses is related, from a configurational standpoint, to D-glyceraldehyde.
- T F (6) Acid halides are extremely reactive compounds.
- T F (7) A sodium lamp generates plane-polarized light.
- T F (8) The specific rotation of a compound depends on the number of molecules of the optically active materials that are in the path of the plane-polarized light.
- T F (9) All molecules that contains only one asymmetric carbon are chiral, but all chiral molecules do not contain an asymmetric carbon atom.
- T $\,$ F (10) The coupling constant J is dependent on the spectrometer frequency.

二填空題 (每題)分, 答錯例和 0.25分)

Fill in the blank with the correct word(s) provided at the end of each question.

- (11) Electromagnetic radiation in the ultraviolet region is of energy than electromagnetic radiation in the infrared region. (higher, lower, similar, unknown).
- (12) The $\sigma + \sigma^*$ electronic transitions are at wavelengths than $n + \sigma^*$ transitions. (longer, similar, unknown, shorter)
- (13) An increase in shielding leads to a _____ shift in the nmr spectrum. (upfield, downfield, unchanged, unpredictable)
- (14) Stereoisomers which are mirror images are called (diastereomers, enantiomorphs, anomers, epimers, chiral certers)
- (15) The only aldehyde that can be prepared by hydration of alkynes is . (formaldehyde, acetaldehyde, propaldehyde, glyceraldehyde, butyraldehyde)
- (16) Optically inactive mixture of equal amounts of enantiomorphs is called
 . (racemate, meso compound, conformers,
 structural isomers, solutes)
- (17) Stereoisomers which differ in the configuration of the hemiacetal carbon of the carbohydrate are called (mutarotators, anomers, conformers, epimers, optical antipodes)
- (18) Separation of a racemic mixture into its optically active forms is called . (purification, resolution, racemization, elimination, identification)
- (19) The group least likely to be listed as an auxochrome is (a) -OH (b) -CH $_3$ (c) -OCH $_3$ (d) -NH $_2$ (e) -NH(CH $_3$)
- (20) The group that has its stretching vibration occurred at the highest frequency is (a) C= C (b) C= C (c) C= 0 (d) C-C (e) C-0 230

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三.寫出化學構造(每題二分)

Write the correct structural formulas for each of the following compounds.

- (21) N-Methylpropanamide
- (22) 2-Mercaptoethanol
- (23) Diisopropyl disulfide
- (24) Aspirin
- (25) Tripeptide glycylphenylalanylalanine

四簡答題及配合題

- (26) Arrange in decreasing acidity. (3%)
 - (a)

CO2

(b)

HO CO7H

(c)

COAH

- (27) Arrange in order of acidity (greatest to least): (3%
 - (a)

(c)

- CN COST
- CH2 CO2H

- (28) (5%) of the following compounds,
 - (a)
 - но
- (h)
- OH OHO OHO
- HO HO HO

- (d)
- R-E-N-E-E-S-CH
- (e
 - C = CH -CH CH20r
- (f)

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	an alkaloid.
	a tetracycline.
	a fat-soluble vitamin.
	a steroid.
	a penicillin.
(29)	(5%) of the following compounds:
	(a) O (b) O (c)
	(d) (e) (f) $(H_{3}-0-\overset{\circ}{C}-((H_{3})_{7}-CH=CH((H_{3})_{7}-CH_{3}) \\ CH-0-\overset{\circ}{C}-((H_{3})_{7}-CH=CH((H_{3})_{7}-CH_{3}) \\ CH_{3}-0-\overset{\circ}{C}-((H_{3})_{7}-CH=CH((H_{3})_{7}-CH_{3}) \\ CH_{3}-0-\overset{\circ}{C}-((H_{3})_{7}-CH_{3}) \\ CH_{3}-0-\overset{\circ}{C}-((H_{3})_{7}-(H_{3})_$
	an oil.
	a wax.
•	a lecithin.
-	a prostaglandin. a phosphatidic acid.
(30)	
	unknown A — positive
	I ₂ , NaOH
	CHI ₃ + others
WI	nich of the following could be the unknown A?
. (1	(b) (c) (d) (e) (e) (H ₃ -C-CH ₃ (H ₃ -C-CH ₃)

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- (31) (2%) Which is more acidic? Why?
 - (a)

(b)

- ozN +Bu
- (32) (3%) Explain why the pKa for phthalic acid, as shown in formula (a), is smaller than the pKa for terephthalic acid, as shown in formula (b).
 - (a) (b) (b) (co2H) HO3(

- (33) (4%) Explain why pyrrole readily reacts with electrophiles at the beta position, whereas pyridine does not.
- (34) (5%) How would you separate a mixture of the following? Assume that they are dissolved in diethylether.
 - ()-CO2H , ()-OH , ()
- (35) (3%) Suggest chemical tests that would distinguish among the following:

$$\bigcirc^{NH_2}, \bigcirc^{NH-CH_3}, \bigcirc^{N-CH_3}$$

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- (36) (3%) Identify the configuration of each of the following as R or S:
 - (a) CH₃
- (b) CD2H
- (C) CO₂H
- (37) (3%) Identify each of the following as E or Z:
 - (a) CH₃ = C'
- (b) CH, H
- (c) H c = c H
- (38) (2%) of the following solvents: (a) CCl₄ (b) D_2O (c) CH_3CH_2OH (d) $CDCl_3$ (e) CD_3COCD_3 , the least desirable for use in nmr spectroscopy is ______.
- (39) (1%) The longer the wavelength (λ) of light the the energy. (greater, less)
- (40) (2%) Esters of long-chain, unbranched fatty acids and long-chain alcohols are called _____.
- (41) (2%) How many sets of nonequivalent protons are there in
 - (a) (1-(H2-(H-(H2-(H3 (H3
- (b) H C = c/H
 H 'C = c/H
- (42) (2%) Given the pKb of $NH_3 = 4.76$, of $CH_3NH_2 = 3.38$. Which is the stronger base?
- 五. Complete the following reactions: (每題三分)
 - $(43) \qquad (H_3-cH_2-cH_2-cO_2H) \xrightarrow{P} \qquad (a)$

1) KOH (b)

(44)

$$CH_3 - C - CH_3 - CH \xrightarrow{CH_3 - CH_3 - CH_3} CH_3 \xrightarrow{CH_3 - CH_3 - CH_3} CH_3 \xrightarrow{CH_3 - CH_3 - CH_3} CA)$$

$$CH_3 - C - CH_3 - CH_3$$

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- (45) CH3 conc. HNO3
 tuming H2 SO4
 ref lux
- $(46) \qquad \qquad \stackrel{0}{\longleftrightarrow} \qquad \frac{\phi_{j} \stackrel{\circ}{p} c_{H_{3}} \stackrel{\circ}{B_{r}}}{(c_{H_{3}})_{j} c_{O} N_{a}}$
- 六. Propose plausible mechanism, (英段四分)

 (47) OFT HO(H,CHOH) OFT