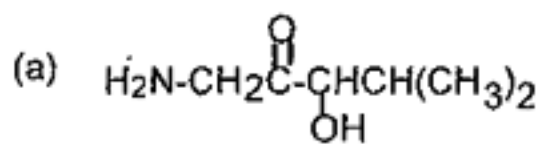
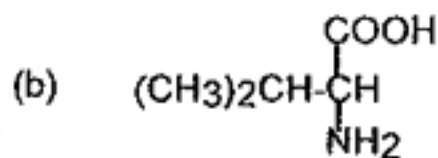
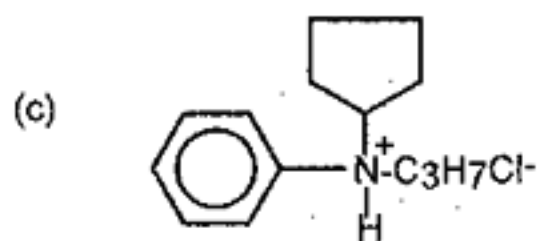
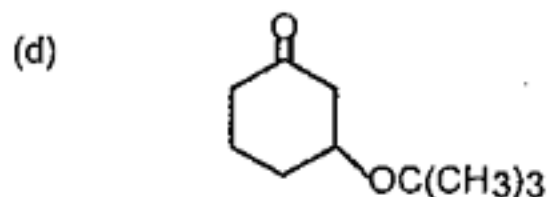


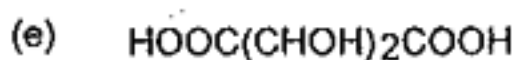
1. What are the IUPAC names of the following compounds (2% for each)?

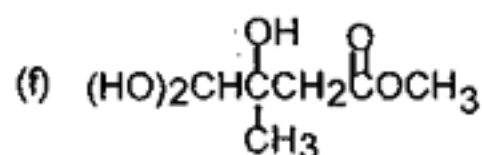


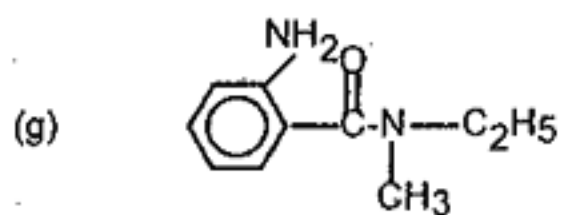


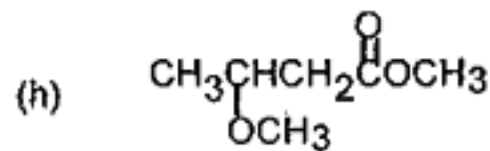


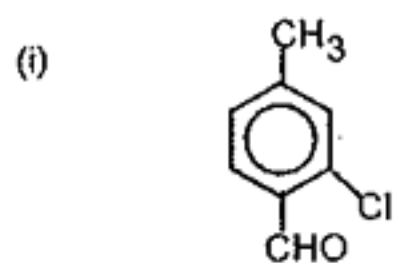


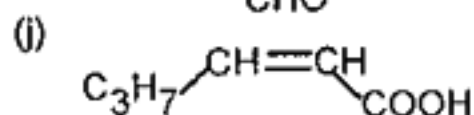




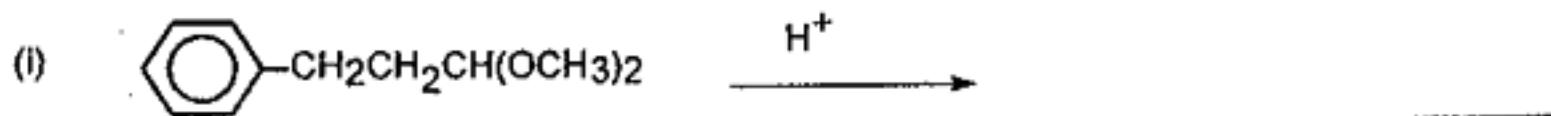
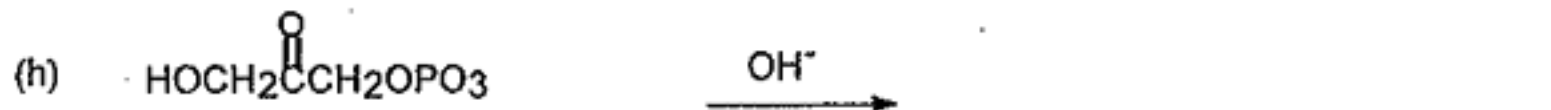
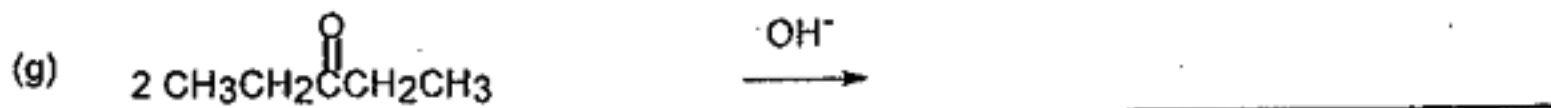
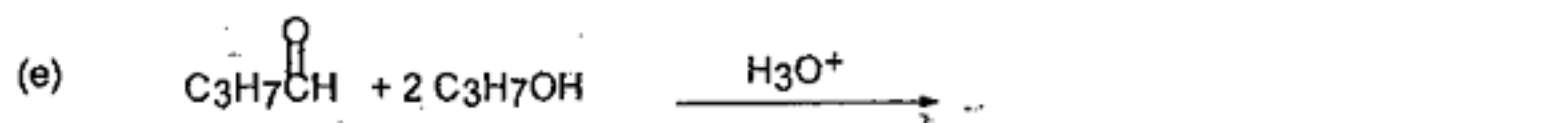
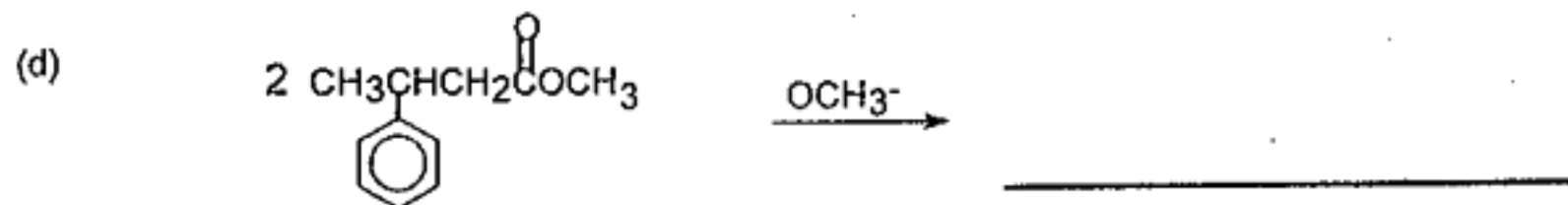
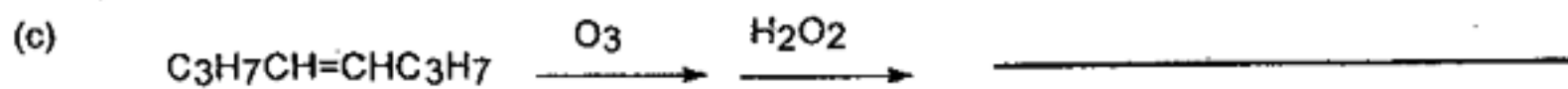
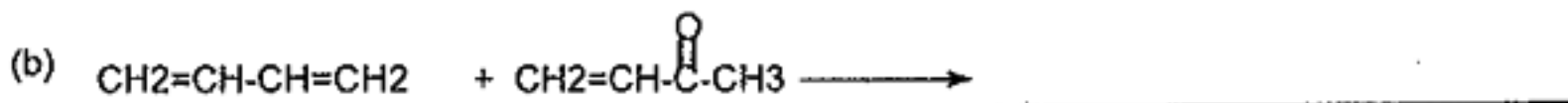
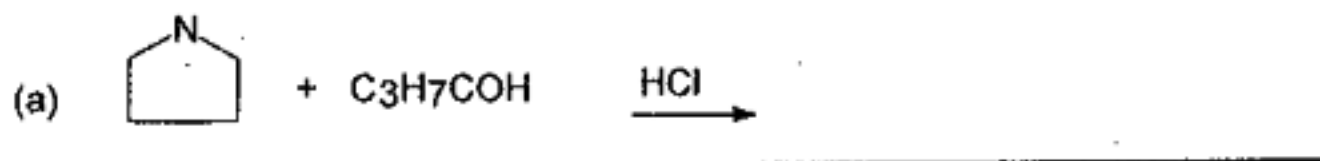




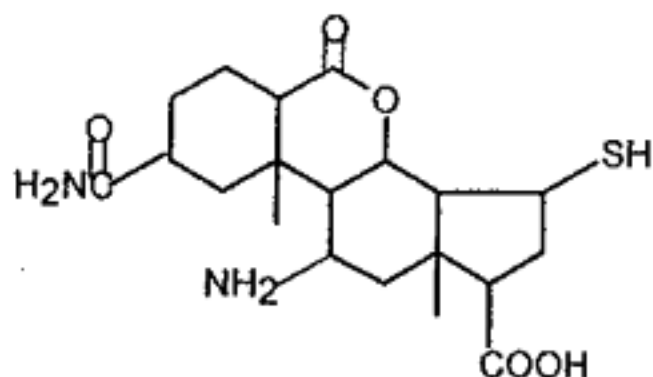




2. Write the structures of products of the following reactions (2% for each).



3. Please identify the functional groups of the following compound and write down the name of functional groups (5%).



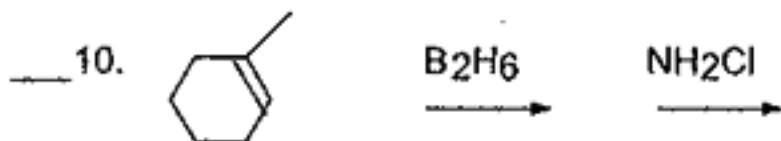
Answer the following questions (2% for each; minus 0.5% for wrong answer)

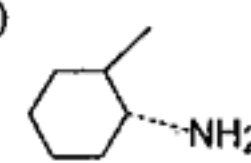
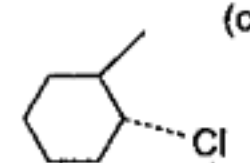
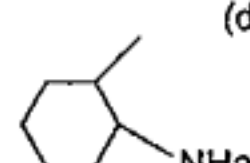
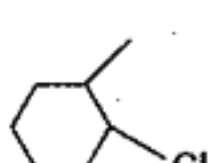
- ___ 4. Which of the following system does not exist as buffer in human body?
(a) Phosphate system
(b) Carbonate system
(c) Protein system
(d) NADH system
- ___ 5. The products of basic hydrolysis of an ester are
(a) another ester + water
(b) alcohol + acid
(c) acid + water
(d) carboxylate salt + alcohol
- ___ 6. The ion formed from a carboxylic acid is called the
(a) carboxylate anion
(b) carboxylate cation
(c) ester
(d) amide
- ___ 7. Asthma would cause?
(a) CO_2 to increase, hence increased pH.
(b) CO_2 to increase, hence decreased pH.
(c) CO_2 to decrease, hence increased pH.
(d) CO_2 to decrease, hence decreased pH.
- ___ 8. What ester would you need to produce $\text{C}_3\text{H}_7\overset{\text{O}}{\parallel}\text{CCH}(\text{C}_2\text{H}_5)\overset{\text{O}}{\parallel}\text{COC}_2\text{H}_5$?
(a) $\text{C}_3\text{H}_7\text{COOCH}_3$
(b) $\text{C}_3\text{H}_7\text{COOC}_2\text{H}_5$
(c) $\text{C}_2\text{H}_5\text{COOC}_2\text{H}_5$
(d) $\text{C}_3\text{H}_7\text{COOC}_3\text{H}_7$

(背面仍有題目,請繼續作答)

9. Which of the following compounds gives a highest pKa?

- (a) $C_6H_5NH_2$
- (b) $C_6H_5SO_2NH_2$
- (c) $C_6H_5CONH_2$
- (d) $CH_3(CH_2)_5NH_2$



- (a) 
- (b) 
- (c) 
- (d) 

11. Which of the following would you not expect to be a reaction of benzene?

- (a) chlorination
- (b) hydrogenation
- (c) sulfonation
- (d) nitration

12. Oxidation of a secondary alcohol will yield a(n):

- (a) aldehyde
- (b) ketone
- (c) carboxylic acid
- (d) no reaction

13. Which molecule shown is heterocyclic?

- (a) aniline
- (b) phenol
- (c) benzoic acid
- (c) pyridine

14. What product is obtained from the oxidation of CH_3CH_2SH ?

- (a) CH_3COH
 $\begin{array}{c} || \\ S \end{array}$
- (b) $CH_3CH_2-S-CH_2-CH_3$
- (c) $CH_3CH_2-S-S-CH_2-CH_3$
- (d) $CH_3CH_2-S-S-H$

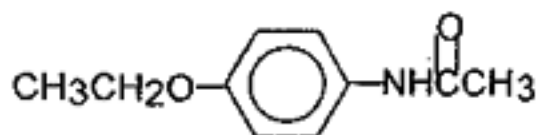
15. In NMR spectrum, the largest downfield shift will be exhibited by the proton or protons in which compound?

- (a) $R-CH_3$
- (b) $R-CH_3O$
- (c) $R-CH_2-Cl$
- (d) $R-COOH$

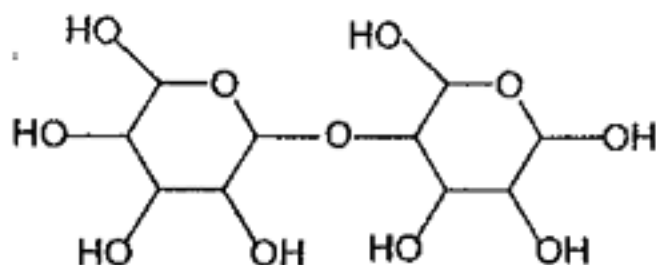
16. What is the intermediate involving aldol condensation?

- (a) carbocation
- (b) radical
- (c) enol
- (d) carboanion

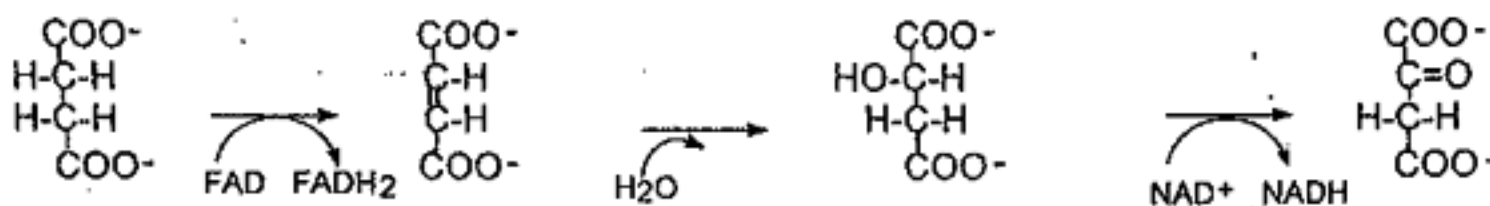
17. Show how phenacetin, a drug once used in headache remedies, can be prepared from the reaction of an anhydride and amine (5%).



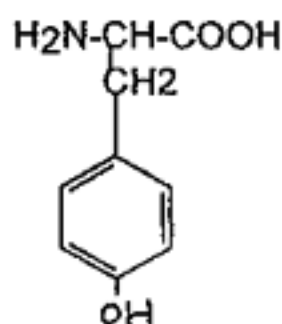
18. What is the results of complete acid hydrolysis of maltose (5%)?



19. (1) Categorize each of the following reactions as addition, oxidation, or elimination.
 (2) Describe in words the chemical changes that take place in the following series of biochemical change (6%).



20. Please sketch the spectrum of tyrosine in D₂O and explain its chemical shift, scalar coupling, and intensity (6%).



21. Propose a structure for the compound based on the UV, IR, Mass, and NMR spectra and explain what kinds of structural informations from these spectra (7%) ?

