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國立成功大學九十八學年度碩士班招生考試試題

共 / / 頁,第 / 頁

生物化學暨分子生物學研究所甲、乙組

考試科目:

有機化學

考試日期:0308,節次:2

☑不可 使用計算機 ※ 考生請注意:本試題 □可

1-45 (2 points for each; wrong answer will deduct 1 point)

1-3. Consider the reaction below to answer the following 1-3 questions.

The nucleophile in this reaction is?

Α Α.

В В.

C. C D. D

E. None of them

2. The electrophile in this reaction is?

A. Α В. В

C. \mathbf{C} D. D

E. None of them

3. The product that results from 1,4-addition is?

Α.

В

C.

D. D

C and D E.

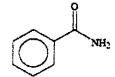
4. Rank the following compounds from most (1) to least (4) acidic.

A. 1>2>3>4 В. 3>2>1>4

C. 1>2>4>3 D. 3>2>4>1

E. 2>4>1>3

5. Which of these gives mostly the meta product when treated with Br2/Fe?



b.

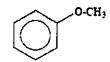




d.



e.



(背面仍有題目,請繼續作答)

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6. Which undergoes electrophilic substitution on the ring most rapidly?

a.

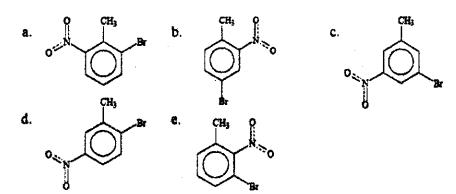
b. C1

c. SO₃H

d. OH

е. СНО

- 7. Which of the following would be the most likely to undergo a nucleophilic aromatic substitution with hydroxide ion in normal conditions?
- a. Benzene
- b. Chlorobenzene
- c. Benzoic acid
- d. p-Chlorotoluene
- e. 2,4,6-Trinitro-1-chlorobenzene
- 8. In the reaction of 2-nitrotoluene with bromine in the presence of iron, which of the products shown below is the most abundant in the mixture?



- 9. Which compound will reduce C=O but not C=C (at least not much)?
 - A. Jones' Reagent
 - B. PCC
 - C. LiAlH4
 - D. NaBH4
 - E. Chromic Acid
- 10. Alkoxymercuration followed by borohydride reduction would be used to produce
 - A. an alcohol from an alkene.
 - B. an aldehyde from alcohol.
 - C. an acid from and alkyne.
 - D. an ether from an alkene.
 - E. an alkene from an aryl halide.

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- 11. Which of the following is not a common reaction of aldehydes?
 - A. Nucleophilic acyl addition
 - B. Nucleophilic acyl substitution
 - C. alpha Substitution
 - D. Carbonyl condensation
 - E. Reduction to a ketone
- 12. In proton NMR, which compound or groups will show a characteristic peak near 10 ppm??
 - A. Alcohols
 - B. Aldehydes
 - C. Ketones
 - D. C=O
 - E. Methyl on a carbonyl
- 13. Hemiacetals and acetals are often found in
 - A. carbohydrates.
 - B. proteins.
 - C. fats.
 - D. oils.
 - E. DNA.
- 14. What would be the major product of the following reaction at 100°C?

A.I

B.II

C.III

D.IV

E.V

- 15. Which is the only one of these compounds which cannot self-condense in the presence of dilute aqueous alkali?
 - A. Phenylethanal
 - B. Propanal
 - C. 2-Methylpropanal
 - D. 3-Methylpentanal
 - E. 2,2-Dimethylpropanal

(背面仍有題目.請繼續作答)

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16. Choose the reagent(s) that would bring about the following reaction:

CH3CH2CH2COOH — CH3CH2CH2CH2OH

- A. H₂/Ni
- B. Li/liq NH₃
- C. LiAIH[OC(CH₃)₃]₃
- D. NaBH₄, CH₃OH
- E. LiAlH₄, ether
- 17. The IR spectrum of a compound exhibits a broad absorption band at 2500-3000 cm⁻¹ and a sharp band at 1710 cm-1. Which of these compounds could it be?
 - A.1-Butanol
 - B. Propyl acetate
 - C. Butanoic acid
 - D. Acetyl chloride
 - E. Acetic anhydride
- 18. What is the product of the reaction below?

D.IV

E.V

19. What is the structure for J?

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Which reagent would best serve as the basis for a simple chemical test to distinguish

CH3CH2CCH2CH3 and CH3CH2CH2CCH3 between

- A. NaOI (I₂ in NaOH)
- B. Br₂/CCl₄

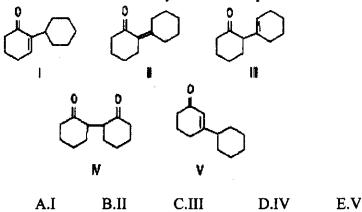
C. CrO₃/H₂SO₄ D. NaHCO₃/H₂O

E. $Ag(NH_3)_2^{\dagger}$

21.

CH3CCH2CH2CH2C=0 produces which of these? The aldol cyclization of

22. The aldol reaction of cyclohexanone produces which of these self-condensation products?



Which reagent would best serve as the basis for a simple chemical test to distinguish between 23.

- NaOI (I₂ in NaOH) Α.
- В. Br₂/CCl₄

C. CrO₃/H₂SO₄

NaHCO₃/H₂O D.

E. $Ag(NH_3)_2^{\dagger}$

(背面仍有題目.請繼續作答)

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- 24. In a solution of acetic acid (pKa=4.74) adjusted to a pH of 2.74,
 - A. the ratio of acetate to acetic acid is 10 to 1.
 - B. the ratio of acetate to acetic acid is 100 to 1.
 - C. the ratio of acetate to acetic acid is 1000 to 1.
 - D. the ratio of acetate to acetic acid is 1 to 10.
 - E. the ratio of acetate to acetic acid is 1 to 100.
- 25. Which compound could be subjected to a haloform reaction to produce m-chlorobenzoic acid?

26. Predict the major organic product of the reaction sequence below

27. What is the starting material in the reaction below?

D. B and C

E. none of the above

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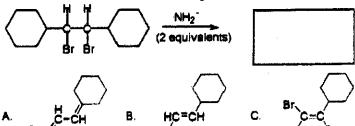
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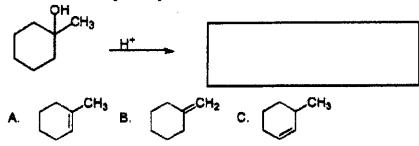
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28. In the reaction below, the product would be



E. none of the above

29. What is the expected product?



D. A and B

E. A. B and C

30. Which of the labeled protons in compound A is most readily abstracted under conditions of E2 elimination?

A. Ha B. Hb C. Hc

D. Hd E. He

31. Which of the following halides is most reactive in an elimination reaction having first order kinetics?

$$A_{r}$$
 B_{r}
 C_{r}
 $CH_{2}B_{r}$
 $CH_{2}I$
 CH_{3}
 CH_{3}
 CH_{3}
 CH_{3}
 CH_{3}

(背面仍有題目,請繼續作答)

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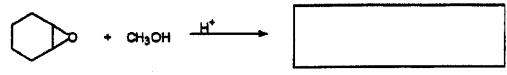
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32. Determine the product.

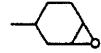


E. none of the above

33. What is the product?

E. none of the above

34. Give the IUPAC name for the following compound?

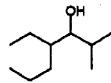


- A. 1,2-epoxy-4-methylcyclohexane
- C. 1,2-epoxy-3-methylcyclohexane
- none of them E.

- B. 3,4-epoxy-1-methylcyclohexane
- D. None of the choices are correct.

35. The reaction below would go via:

- A. SN₁ mechanism with inversion
- C. SN₂ mechanism with stereoretention
- E. SN₂ mechanism with inversion
- B. SN₁ mechanism with racemic product
- D. SN₂ mechanism with racemic product
- 36. Give the IUPAC name for the following compound.



- A. 3-ethyl-2-methyl-4-heptanol
- B. 4-ethyl-6-methyl-5-heptanol
- C. 4-ethyl-2-methyl-3-octanol
- E. None of the choices are correct.
- D. 4-ethyl-2-methyl-3-heptanol

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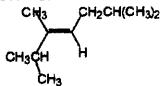
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37. Give the IUPAC name for the following compound.

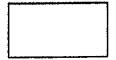


- A. (Z)-2,3,6-trimethyl-2-heptene
- C. (E)-2,3,6-trimethyl-3-heptene
- E. (E)-2,4,6-trimethyl-3-heptene
- B. (Z)-2,3,6-trimethyl-3-heptene
- D. (E)-2,3,6-trimethyl-2-heptene

38. Predict the product.



1. BH₃ 2. H₂O₂/OH



A. OH

в. ОН

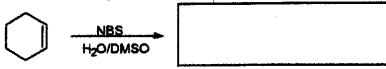
c. OH

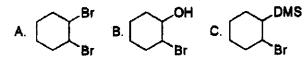
D. OH

E. A and B

- 39. A compound M with molecular formula C₈H₁₂ and no triple bonds reacts with H₂ to give a new compound having molecular formula C₈H₁₄. What can be said about M?
 - A. M has 3 rings.

- B. M has 2 rings and 1 π bond.
- C. M has 1 ring and 2π bonds.
- D. M has 3π bonds.
- E. It's impossible to say anything about the structure of M given this data.
- 40. Predict the product.





D, none of the above

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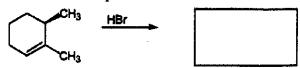
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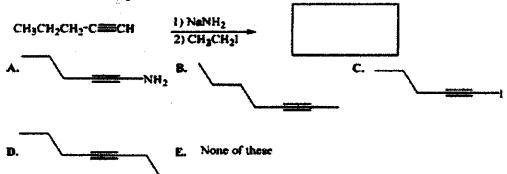
41. Predict the product



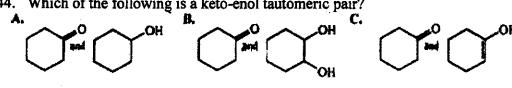
$$A. \qquad \qquad B. \qquad G. \qquad Gr$$

42. Predict the product.

- D. A and B E. A and C
- 43. Determine the product?



44. Which of the following is a keto-enol tautomeric pair?



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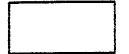
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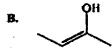
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None of the above

46. Identify the unknown compound that has all the following characteristics. (2 points)

Molecular formula: C₁₅H₁₄O

¹H NMR:

singlet, δ 2.20 (3H)

singlet, δ 5.08 (1H)

multiplet, δ 7.25 (10H)

IR spectrum: strong peak at 1720 cm⁻¹

47. Identify the unknown compound that has all the following characteristics. (2 points)

Molecular formula: C8H9Br

¹H NMR:

doublet, δ 2.0 (3H)

quartet, δ 5.15 (1H)

multiplet, δ 7.35 (5H)

48. What distinguishing features would appear in the IR spectra of the following compounds? (i.e. give approximate absorption range and intensity).

49. Identify the unknown compound that has all the following characteristics. (2 points)

Molecular formula C10H12O.

IR: peak at 1710 cm⁻¹

¹H NMR:

singlet, δ 2.1, 3H

multiplet, δ 3.0, 4H

multiplet, δ 7.1, 5H

50. Which is (are) the least shielded?