編號: 415 國立成功大學九十九學年度碩士班招生考試試題 共 / 頁·第/頁

系所組別: 口腔醫學研究所丙組

考試科目: 普通化學

考試日期:0307·節次:3

(15%)

※ 考生請注意:本試題 □ □ 不可 使用計算機

- Please describe and explain (a) salt bridge, (b) first ionization energy, (c) enthalpy,
 (d) surface energy, (e) specific heat capacity.

 (20%)
 - 2. Describe and compare Ideal Gas and Real Gas. (10%)
 - The periodic table is grouped into "A" and "B". Please describe the distinction between IA and IIA elements. (10%)
 - 4. Calculate the molar solubility of PbF₂ in water? The K_{sp} for PbF₂ is 2.7×10^{-8} (10%)
 - Calculate the ratio of root-mean-square molecular speed of CO₂ and SO₂ at 25°C and 100 atm? (Atomic weight, C = 12, O = 16, S = 32) (10%)
 - Predict whether the following molecules are polar or nonpolar, (a) KCl, (b) CH₄,
 (c) CO₂. You need to explain your prediction. (15%)
 - 7. Describe in detail steps how to prepare 100 mL of 1.0 M $\rm H_2SO_4$ from the

concentrated sulfuric acid (95% w/w, specific gravity 1.83)

A first-order reaction is 30% complete in 480 s. (a) calculate the rate constant, (b)
 What is the value of the half time? (10%)