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系所組別: 口腔醫學研究所丙組 考試科目: 材料科學

※ 考生請注意:本試題 ☑ □ □ 不可 使用計算機

- Please explain why FCC (face-centered cubic) metals are typically more brittle than and BCC (body-centered cubic) metals. (10%)
- Sketch the unit cells for hexagonal close -packed crystal structures, and sketch the (0111) and (2110) planes in a hexagonal unit cell. (10%)
- Describe recrystallization in terms of both the alteration of microstructure and mechanical characteristics of the material.

(10%)

- Compare the mechanism of crack propagation for both ductile and brittle modes of fracture. (10%)
- 5. Why a metal having small grains is stronger than one having large grains? (10%)
- List and briefly describe at least three testing methods of mechanical properties of materials. (10%)
- Calculate the radius of an vanadium atom, given that V has an BCC crystal structure, a density of 5.96 g/cm³, and an atomic weight of 50.9 g/mol. (10%)
- Explain (a) cold working, (b) coordination number, (c) edge dislocation, (d) sintering, (e) plastic deformation. (20%)
- Describe four features of the solute and solvent atoms that from the substitutional type solid solution. (10%)