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I. 單選題 (每題 2 分)

- The equilibrium constant for the ionization of acetic acid, CH₃COOH ← → CH₃COO⁻ + H⁺, is 0.00002. What can you conclude about this reaction?
 - A. It is a closed system
 - B. At equilibrium, the concentration of CH₃COOH is much higher than the concentrations of CH₃COO⁻ + H⁺
 - C. It never reaches equilibrium
 - D. It is spontaneous, starting with CH3COOH
 - E. The equilibrium constant increases when the starting concentration of CH₃COOH is increased.
- 2. In the presence of alcohol dehydrogenase, the rate of reduction of acetaldehyde to ethanol increases as you increase the concentration of acetaldehyde. Eventually the rate of the reaction reaches a maximum, where further increases in the concentration of acetaldehyde have no effect. Why?
 - A. all of the alcohol dehydrogenase molecules are bound to acetaldehyde molecules
 - B. at high concentrations of acetaldehyde, the activation energy of the reaction decreases
 - C. the enzyme is no longer specific for acetaldehyde
 - at high concentrations of acetaldehyde, the change in free energy of the reaction decrease
- 3. The specific energy source for the reaction ADP + phosphate →ATP by the enzyme ATP synthetase (CF1 Coupling Factor) in thylakoid membranes is:
 - A. oxidation of water
 - B. oxidation of NADPH
 - C. absorption of photons by chloroplast pigments
 - D. higher concentration of H⁺ inside versus outside the thylakoid membranes
- 4. Which is a common site for attachment of carbohydrates in glycoprotein?
 - A. Thr
 - B. Ser
 - C. Gln
 - D. Asn
 - E. Tyr

(背面仍有題目,請繼續作答)

93學年度國立成功大學 生物科技研究所 甲組 生物化學 7/2

- 5. What type compound is similar to RU486?
 - A. peptide
 - B. steroid
 - C. carbohydrate
 - D. nucleotide
- 6. What cell in the Pancreas secreted the insulin?
 - A. α cell
 - B. β cell
 - C. χ cell
 - D. δ cell
- 7. How many ATP consumed that six CO2 molecules fixed in the Calvin cycle?
 - A. 9
 - B. 12
 - C. 15
 - D. 18
- 8. What inorganic element is the cofactor for urease?
 - A. Ni²⁺
 - B. Fe²⁺
 - C. Cu2+
 - D. Zn²⁺
- 9. What are the possible metabolic fates of pyruvate except:
 - A. NADH
 - B. lactate
 - C. ethanol
 - D. acetyl group of acetyl-coenzyme A
- 10. What are the roles of citric acid cycle in organism except:
 - It produces cofactor NADH, FADH₂ and Biotin.
 - B. It produces intermediates as precursors for other biosynthetic pathways.
 - C. It produces energy.
 - D. It occurs in mitochondria only.
 - All reactions are catalyzed by enzymes.
- 11. Oxidation of fatty acids is tightly regulated. What is the rate-limiting step in fatty acid oxidation and is an important point of regulation:

就題 第3頁

- A. Fatty acid is catalyzed to fatty acyl-CoA in cytosol.
- B. Fatty acyl-CoA enters into mitochondria matrix.
- C. Rate of β-oxidation of fatty acyl-CoA in mitochondria matrix.
- D. Acetyl-CoA enter into citric acid cycle.
- E. Rate of FADH₂ and NADH go through the electron transport chain.
- 12. What is the common process between oxidative phosphorylation and photophosphorylation:
 - Both need water as electron donor
 - B. Both need oxygen as electron acceptor
 - C. Both have identical electron transport chain
 - D. Both processes can produce energy
 - E. Both process need light to activate the reactions.
- 13. Which amino acid is not belongs to aromatic R group.
 - A. Phenylalanine
 - B. Tyrosine
 - C. Tryptophan
 - D. Lysine
- 14. How many levels of protein structure.
 - A. 1
 - B. 2
 - C. 3
 - D. 4
- 15. Which substrate could be recognized by Ribozyme.
 - A. RNA
 - B. DNA
 - C. Lipid
 - D. Protein

II. 複選題 (毎題 2 分)

 During a heart attack, blood flowing to the heart muscle is interrupted by blockage of a coronary artery. How would you expect the metabolism in the heart to change?

(背面仍有题目,請繼續作答)

93學年度國立成功大學 生物科技研究所 甲組 生物化學

試題 共 6 頁

- (a) oxidative phosphorylation would slow down in the mitochondria
- (b) the rate of production of lactic acid would be stimulated
- (c) the production of water by mitochondria would be inhibited
- (d) the use of glucose by the muscle tissue would increase
- Which of the following are weak forces associated with the interaction of biological molecules
 - (a) hydrogen bonds
 - (b) electrostatic attractions
 - (c) covalent bonds
 - (d) Van der Waals interactions
- 3. What organelles involved to the conversion of fatty acid to sucrose in germinating seeds?
 - A. Glyoxysome
 - B. Mitochondria
 - C. Chloroplast
 - D. Cytosol
- 4. What inorganic element is the cofactor for cytochrome oxidase?
 - A. Zn^{2+}
 - B. Cu²⁺
 - C. Fe²⁺
 - D. Ni²⁺
- 5. How the activity of urea cycle is regulated?
 - A. High protein diet increase urea production
 - B. High carbohydrate diet increase urea production
 - C. Prolonged starvation increase urea production
 - Prolonged starvation decrease urea production
 - E. N-acetylglutamate activate the enzymes in urea cycle
- 6. What are the correct description regarding hemoglobin:
 - A. It is a tetrameric protein.
 - B. 2.3-Bisphosphoglycerate can regulate the binding of oxygen to hemoglobin.
 - C. Structural change of hemoglobin can affect the binding of oxygen.
 - D. Sickle-cell anemia is a disease caused by the point mutation of hemoglobin.
 - E. It is responsible for transporting oxygen only.

- 7. What are the correct description regarding pyruvate dehydrogenase complex?
 - A. It proceeds with three different enzyme reactions in sequential way.
 - B. It require 5 coenzymes such as TPP, FAD, CoA, NAD and lipoate
 - C. The substrate of enzyme reaction are pyruvate and CO2
 - E. The product of enzyme reaction is lactate
 - F. The product of enzyme reaction is acetyl-CoA
- Which codon is belongs to termination codon normally signal the end of polypeptide synthesis.
 - A. UAA
 - B. UAG
 - B. UGA
 - C. UUU
- 9. How many codons could code for glycine
 - A. 1
 - B. 2
 - C. 3
 - D. 4
- 10. Which restriction endonuclease is belongs to sticky end enzyme.
 - A. BamHI
 - B.ClaI
 - C. EcoRI
 - D. HaeIII

III. 簡答題 (每題 5 分)

- 1. Branched-chain fatty acids are found in some bacterial membrane lipids. Would their presence increase or decrease the fluidity of the membrane (that is, make them have a lower or higher melting point)? Why?
- 2. What factors that affect the melting point (Tm) of DNA?

- 3. Which of the following graphs shows the results of reaction rate vs substrate concentration for an non-allosteric enzyme in the absence and presence of a non-competitive inhibitor (non-competitive inhibitors bind to an enzyme at a site different than the active site)? Explain
- 4. If linking number (Lk) of the supercoil plasmid is 200 and the twice writhe (Wr) for the plasmid that has the positive supercoil. Please answer how many is the twist (Tw).
- 5. The X protein isolated from mouse brain has 415 amino acid residues and is coded by a cDNA with 2,850 base pairs. Explain the structure of cDNA for X protein:
- 6. What are the chemiosmotic model? How it can explain the ATP synthesis in oxidative phosphorylation?
- 7. Why ketone bodies are overproduced in liver of untreated diabetes patient and/or during starvation?
- 8. Please described a function gene how to expression from transcription to translation level.
- 9. How do you clone a growth hormone gene from a cDNA library when you got a specific probe from fish system?
- 10. What is functional genomics? And how to apply on new drug discovery?