编號:	49	國立成功大學一〇〇學年度碩士明	田招生考试试题 共4頁	,第 頁
系所組別	: 化學系		·	,
考試科目	: 無機化學	/	考試日期:0219	,節次: 3
※ 考生語	青注意:本試題 🗌	可 以不可使用計算機		
諸勿在 一選	本試題紙上作答, 擇題 (每題4分)	否則不予計分		
1. For interact π inter	a six-coordinate contribution between d_{xz} orbitation, (d) δ interaction, (d)	nplex with an octahedral geometry such pital of metal and p _x orbital of ligand? tion)	n as $M(L)_6$ complex, what is likely (a) no interaction, (b) σ interaction	ı, (c)

2. What is the order of acidity in gas phase? (a) $SiH_4 < PH_3 < H_2S < HCl$ (b) $SiH_4 > PH_3 > H_2S > HCl$ (c) $PH_3 < H_2S < HCl < SiH_4$ (d) $H_2S < HCl < SiH_4 < PH_3$

3. Please order the bond dissociation energy for the following species C_2^+ , C_2 , C_2^- (a) $C_2^- > C_2 > C_2^+$ (b) $C_2^+ > C_2 > C_2^-$ (c) B) $C_2^+ = C_2 = C_2^-$ (d) $C_2 > C_2^- > C_2^+$

4. Determine the chirality label for the complex on the right.

(a) λ (b) δ (c) Δ (d) Λ



0 0 = oxalate

5. Which of the following statement for compound I and compound II shown on the right figure is incorrect?

a) compound I is a facial form

b) compound II is a meridional form

c) compounds I and II are geometric isomers

d) compound I and II are optical isomers



Compoun I

Compoun II

6. Which of the following compound is most likely to stabilize dihydrogen species rather than dihydride species?

(a) $Mo(PMe_3)_5H_2$ (b) $Mo(PF_3)_5H_2$ (c) $Mo(PMe_3)_4(CO)H_2$ (d) $Mo(PF_3)_4(CO)H_2$

7. Which of the following metalloproteins does not play the role for electron transfer?
(a) iron sulfur protein
(b) coenzyme B₁₂
(c) cytochrome a
(d) blue copper protein

8. Which metal shown below has the most affinity for binding to the amino acid, methionine?,
(a) Cu²⁺, (b) Co³⁺, (c) Hg²⁺ (d) Fe²⁺

(背面仍有題目,請繼續作答)



10. Considering a ML₆ complex, how many of the following complexes have strong Jahen-Teller effects? high-spin d³ complex, low-spin d³ complex, high-spin d⁴ complex, high spin d⁵ complex, low-spin d⁷ complex, d⁹-complex
(a) 6 (b) 5 (c) 4 (d) 3 (e) 2

11. What is the number of atoms in each unit cell of a faced centered cubic structure?a) 1b) 2c) 3d) 4

12. Which of the following is not the Lattice point of a faced centered cubic structure a) (0,0,0) b) (1/2, 1/2, 0) c) (1/2, 1/2, 1/2) (d) (1/2, 0, 1/2) e) (0, 1/2, 1/2)

13. HA and B has a very strong hydrogen bonding.Which one is the possible MO diagram for the frontier orbitals of HA and B? (a), (b), or (c)?

14. Please predict the order of the bond distance for different CO (a, b, and c) in compound X on the right. (a)
a>c>b (b) b>c>a (c) a>b>c (d) c>a>b



編號: 49 系所組別: (1 考試科日: 4	上學系	國立成功大學	┣━○○學年)	度碩士班招生考試試題	共4頁,	€ 第 3 頁				
					考試日期:0219,	節次:3				
※考生請注注	意:本試題 □□□	可 ☑不可 使	用計算機							
15. determin (a) 20 (b	ne the valance ele b) 19 (c) 18	ectron count fo (d) 17 (e)	r the transition 16	metal in [(η ⁵ -C ₅ H ₅)(cis-r	4 ⁴ -C4H6)Fe(PMe3)()	H)]				
二 簡答題	(每題4分)									
<u>Considering</u>	<u>the character to</u>	able of C _{3v} and	<u>answer the foll</u>	owing three questions.						
$C_{3 u}$	E	$3\sigma_{\nu}$	2C3			ļ				
A_1	1	1	1							
A2	1	-1	1	Rz		i				
Ε	1			(x,y), (Rx,Ry)		ľ				
 What are What is the How man 	characters for E he irreducible rep by bands do you	representation? presentation for expect in IR sp	z coordinate in ectrum for NH	this point group? molecule?						
4. Please ind	dicate the point g	group of cycloh	exane (chair co	nformation)						
5. How many microstates for an s ¹ p ¹ configuration?										
6. Determin	e the ground terr	ns for low-spin	d ⁷ configuration	ons in O _h symmetry.						
The electror	<u>uic spectrum of [</u>	<u>M(H2O)6]ⁿ⁺ an</u>	<u>d the simplified</u>	<u> Tanabe-Sugano Diagrai</u>	<u>ms in octahedral_</u>					
<u>ligand field</u>	are shown in the	e appendixP	lease answer th	e following two question	<u>s.</u>					
7. Please in	ndicate the term	of the ground s	tate for [Cr(H ₂	$(D)_{6}]^{2+}$						
8. Please indicate the term of the ground state for $[Fe(CN)_6]^{4-1}$										
9. Please dr	aw the ligand ac	etylacetonato (a	acac).			and the second				
10. Please draw the d-orbitals splitting diagram for a five-coordinate complex with trigonal bipyramidal geometry.										

(背面仍有題目,請繼續作答)

