

(1) A charged particle with mass  $m$  and charge  $+q$  is released from rest in an uniform electromagnetic field with  $\vec{E} = E_0 \hat{e}_y$ ,  $\vec{B} = B_0 \hat{e}_z$

12%

- (a) Express the forces exerted on the particle while the particle is in motion.
- (b) What are the equations of motion of the particle?
- (c) Verify that the motion of the particle will be limited on the  $x$ - $y$  plane?

(2) Using the results of Kepler's laws of planetary motion, show that

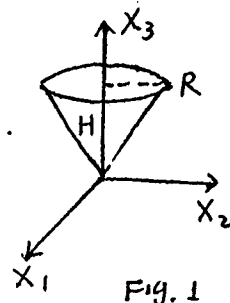
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- (a) the gravitational force must be central.
- (b) its radial dependence must be  $\frac{1}{r^2}$ .

(3) Consider a symmetrical top with radius  $R$ , height  $H$ , and mass  $m$  as shown in Fig. 1

14%

- (a) Find the position of the center of mass.
- (b) Find the moment of inertia of the top about  $X_3$ -axis.
- (c) If the top spins about the  $X_3$ -axis and makes a steady precession about vertical line with a constant angle  $\theta$ , suppose the precession angular velocity is  $\Omega$ , find the angular velocity of spinning  $\omega$ .



(4) A sphere of radius  $p$  is constrained to roll without slipping on the lower half of the inner surface of a hollow cylinder of inside radius  $R$ . Determine the Lagrangian function, the equation of constrain, and the Lagrange's equations of motion. Find the frequency of small oscillation

12%

(5) Prove that (a) the change in enthalpy during an isobaric process is equal to the heat that is transferred (b) in the case of a reversibly isothermal and isobaric process, the Gibbs free energy remains constant

15%

(6) (a) What is called the Carnot Cycle?

15% (b) Show that the maximum efficiency of any engine operating between two reservoirs is that of a Carnot engine operating between the same two reservoirs.

(7) A system consists of  $N_A$  distinguishable, independent particles, each of which is capable of existing in only two nondegenerate energy states  
20% 0 and  $E$ .

(a) What is the partition function?

(b) Calculate the energy.

(c) Calculate  $C_V$ .

, where  $N_A$  is the Avogadro's number and  $C_V$  is the heat capacity per mole.